

The infrared emission spectra of LiF and HF

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This paper is dedicated to Dr. Richard Norman Jones

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The high resolution infrared spectrum of LiF has been measured in emission with the McMath Fourier transform interferometer at Kitt Peak. A total of 800 lines with $v = 1 \rightarrow 0$ to $v = 8 \rightarrow 7$ of the main isotopomer, ^7LiF , and 250 lines with $v = 1 \rightarrow 0$ to $v = 3 \rightarrow 2$ of the minor isotopomer, ^6LiF , were observed. These ro-vibrational transitions and the pure rotational transitions from the literature were fit to a set of Dunham coefficients Y_{ij} and a set of mass-reduced Dunham coefficients U_{ij} . The same spectrum shows 13 pure rotational emission transitions of HF in the vibrational ground state with $J = 13 \rightarrow 12$ to $J = 25 \rightarrow 24$. These transitions were used to determine an improved set of rotational constants for HF.

Key words: infrared spectra, LiF, HF.

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On a mesuré les spectres d'émission infrarouge à haute résolution du LiF à l'aide d'un interféromètre à transformation de McMath Fourier au niveau Kitt Peak. On a observé 800 raies avec des valeurs de v allant de $v = 1 \rightarrow 0$ à $v = 8 \rightarrow 7$ pour l'isotopomère principal, ^7LiF , et on a observé 250 raies avec des valeurs de v allant de $v = 1 \rightarrow 0$ à $v = 3 \rightarrow 2$ pour l'isotopomère minoritaire. On a adapté ces transitions de vibration-ro et les transitions de rotation pure répertoriées dans la littérature à un ensemble de coefficients de Dunham Y_{ij} et à un ensemble de coefficients U_{ij} de masse réduite Dunham. Ces mêmes spectres montrent 13 transitions d'émission de rotation pure du HF dans l'état fondamental de vibration avec $J = 13 \rightarrow 12$ à $J = 25 \rightarrow 24$. On a utilisé ces transitions pour déterminer un ensemble amélioré de constantes de rotation pour le HF.

Mots clés : spectres infrarouge, LiF, HF.

[Traduit par la rédaction]

I. Introduction

The infrared spectra of alkali halides have been of interest to spectroscopists for a long time (1, 2). However, high-resolution infrared data for these molecules have only become available after the introduction of diode laser and Fourier transform infrared spectrometers. Nevertheless, the first infrared data on LiF was obtained by Vidale (3) as early as 1959 with a Perkin-Elmer grating spectrometer. He observed the spectrum of LiF both in absorption and emission. However, the resolution was not high enough to obtain an unambiguous rotational assignment and, therefore, the correct band center. Vidale gave three possible assignments. Only after the microwave measurements on ^6LiF by Wharton *et al.* (4) did the correct assignment become apparent. Several additional papers report the infrared spectrum of LiF trapped in low temperature matrices (5–10).

The first high-resolution infrared data on LiF was published by Maki (11). He used a tunable diode laser system for the observation of ro-vibrational transitions with $v = 1 \leftarrow 0$ to $v = 8 \leftarrow 7$ for ^7LiF and ^6LiF . Diode lasers display many gaps in their spectral coverage. Therefore, only a few lines of each vibrational band were accessible. In this paper the complete emission spectra of ^7LiF and ^6LiF between 630 and 1030 cm^{-1} are published (Figs. 1, 2).

HF is an important gas for use as an absolute wavenumber standard. In 1987 Jennings *et al.* (12) published rotational constants for HF calculated from measurements in four different laboratories. Highly accurate data in the far infrared ($J = 1 \leftarrow 0$ to $J = 4 \leftarrow 3$) were obtained with a tunable far-infrared spectrometer at the National Bureau of Standards laboratory in Boulder. Transitions with $J = 5 \leftarrow 4$ to $J = 7 \leftarrow 6$ were measured with a commercial high-resolution Fourier transform

spectrometer at the National Research Council, Ottawa, Canada. Lines with $J = 17$ to $J = 21$ were observed with a vacuum grating spectrograph (13) and lines with $J = 26$ to $J = 32$ with a tunable diode laser spectrometer at the University of Lille, France. In 1988 Jennings and Wells (14) remeasured the $J = 27 \leftarrow 26$ and $J = 33 \leftarrow 32$ transitions by using a tunable diode laser heterodyne spectrometer (15). Except for these two lines, the mid-infrared transitions are relatively inaccurate. Unfortunately the uncertainties in the molecular constants in this publication seem to be a factor of 10 too small. As a by-product of our work with LiF, we publish thirteen rotational transitions of HF in the range 500 to 900 cm^{-1} . A new fit including the far-infrared data and the diode laser measurements provide an improved set of rotational constants for HF.

II. Experimental

The high resolution infrared emission spectra of ^6LiF , ^7LiF , and HF between 500 and 1400 cm^{-1} were observed with the McMath Fourier transform spectrometer at Kitt Peak. The unapodized resolution was 0.0055 cm^{-1} with a liquid helium cooled As:Si detector and a KCl beamsplitter. The upper wavenumber limit was set by a wedged InSb filter while the lower limit was given by the transmission of the KCl beamsplitter and by the detector response. The spectrum of LiF was taken accidentally during an attempt to observe the infrared spectrum of MgF_2 .

Solid MgF_2 was heated to about 1500 K in an alumina tube furnace. The apparatus used for the experiment was described in detail earlier (16). The heating rate was about 5 K/min. The temperature of the furnace was measured with a chromel–alumel thermocouple placed between the heating elements and the alumina tube. Deposition of solid material onto the windows was avoided by pressurizing the cell with 5 Torr of argon. A series of spectra were taken as the furnace heated up and then cooled down. Initially a global was placed behind the tube furnace and its image was focused on the 8 mm aperture of the Fourier transform spectrometer. No absorption spectra were observed, but when the global was shut off (at a temperature of about 1300 K) an emission spectrum was monitored. The inten-

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TABLE 1. Ro-vibrational transitions of ${}^7\text{LiF}$ and ${}^6\text{LiF}$

	J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}	J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}
${}^7\text{LiF}$ $v = 1 \rightarrow 0$	P67	643.22499(50) ^a	-32	P4	883.42268(50)	-16
	P66	647.79186(50)	-14	P3	886.21112(50)	-15
	P65	652.34367(100)	11	P2	888.96043(50)	-16
	P64	656.87691(50)	3	P1	891.67033(50)	-19
	P63	661.39365(50)	5	R0	896.97099(50)	-10
	P62	665.89322(50)	21	R1	899.56097(50)	-19
	P61	670.37521(50)	56	R2	902.11051(50)	-23
	P60	674.83778(50)	-20	R3	904.61936(50)	-17
	P59	679.28310(50)	52	R4	907.08707(50)	-21
	P58	683.70808(50)	60	R5	909.51355(50)	-17
	P57	688.11522(150)	130	R6	911.89840(50)	-18
	P56	692.49965(50)	-11	R7	914.24140(50)	-20
	P55	696.86537(50)	1	R8	916.54237(50)	-16
	P54	701.21006(50)	22	R10	921.01692(50)	-16
	P53	705.53328(50)	5	R11	923.19005(50)	-16
	P52	709.83535(50)	38	R12	925.32007(50)	-17
	P51	714.11486(50)	21	R13	927.40677(50)	-17
	P50	718.37200(50)	12	R14	929.44992(50)	-13
	P49	722.60630(50)	6	R15	931.44921(50)	-14
	P48	726.81744(50)	10	R16	933.40445(50)	-16
	P47	731.00488(50)	10	R17	935.31541(50)	-18
	P46	735.16819(50)	2	R18	937.18186(50)	-20
	P45	739.30720(50)	9	R19	939.00368(50)	-12
	P44	743.42133(50)	12	R20	940.78049(50)	-11
	P43	747.51018(50)	9	R21	942.51208(50)	-16
	P42	751.57345(50)	10	R22	944.19838(50)	-11
	P41	755.61069(50)	7	R23	945.83905(50)	-11
	P40	759.62154(50)	3	R24	947.43398(50)	-5
	P39	763.60582(50)	18	R25	948.98282(50)	-9
	P38	767.56266(50)	1	R26	950.48550(50)	-8
	P37	771.49217(50)	3	R27	851.94180(50)	-7
	P36	775.39382(50)	6	R28	953.35151(50)	-5
	P35	779.26714(50)	1	R29	954.71447(50)	-1
	P34	783.11189(50)	1	R30	956.03041(50)	-3
	P33	786.92762(50)	-4	R31	957.29927(50)	1
	P32	790.71410(50)	1	R32	958.52075(50)	-1
	P31	794.47076(50)	-5	R33	959.69477(50)	1
	P30	798.19745(50)	-3	R34	960.82113(50)	3
	P29	801.89368(50)	-4	R35	961.89967(50)	6
	P28	805.55917(50)	-3	R36	962.93020(50)	8
	P27	809.19351(50)	-5	R37	963.91257(50)	9
	P26	812.79639(50)	-5	R38	964.84670(50)	16
P25	816.36742(50)	-9	R39	965.73172(50)	-42	
P24	819.90638(50)	-4	R40	966.56926(50)	13	
P23	823.41278(50)	-5	R41	967.35751(50)	13	
P22	826.88635(50)	-4	R42	968.09690(50)	16	
P21	830.32668(50)	-10	R43	968.78722(50)	14	
P20	833.73363(50)	-3	R44	969.42838(50)	10	
P19	837.10660(50)	-9	R45	970.02040(50)	20	
P18	840.44548(50)	-8	R46	970.56296(50)	23	
P17	843.74986(50)	-6	R47	971.05594(50)	19	
P16	847.01934(50)	-11	R48	971.49941(50)	27	
P15	850.25377(50)	-7	R49	971.89296(50)	16	
P14	853.45267(50)	-10	R50	972.23690(50)	27	
P13	856.61580(50)	-11	R51	972.53012(50)	-41	
P12	859.74286(50)	-9	R52	972.77459(50)	20	
P11	862.83347(50)	-11	R53	972.96841(50)	26	
P10	865.88735(50)	-14	R54	973.11156(50)	-14	
P9	868.90424(50)	-14	R55	973.20486(50)	-11	
P8	871.88380(50)	-13	R58	973.18288(50)	-9	
P7	874.82572(50)	-13	R59	973.07410(50)	28	
P6	877.73041(50)	57	R60	972.91472(50)	5	
P5	880.59549(50)	-11	R61	972.70458(50)	-28	
			R62	972.44380(100)	-56	

TABLE I (continued)

	J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}	J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}
$v = 2 \rightarrow 1$	R63	972.13208(100)	-103	P4	867.67416(50)	-17
	R64	971.76979(150)	-131	P3	870.42072(50)	-25
	R66	970.88707(1000)	-757	P2	873.12879(50)	-32
	R67	631.22869(50)	-31	P1	875.79864(50)	17
	P66	635.72007(50)	-54	R0	881.01967(50)	-9
	P65	640.19664(50)	12	R1	883.57101(50)	-11
	P64	644.65580(50)	-49	R2	886.08249(50)	-11
	P63	649.09925(50)	-24	R3	888.55365(50)	-28
	P62	653.52568(50)	-2	R4	890.98463(50)	-21
	P61	657.93478(50)	30	R5	893.37488(50)	-19
	P60	662.32515(50)	-27	R6	895.72416(50)	-20
	P59	666.69846(50)	30	R7	898.03226(50)	-19
	P58	671.05195(50)	-10	R8	900.29890(50)	-19
	P57	675.38687(50)	-4	R9	902.52385(50)	-17
	P56	679.70218(50)	-7	R10	904.70683(50)	-17
	P55	683.99744(50)	-21	R11	906.84760(50)	-18
	P54	688.27260(50)	-9	R12	908.94670(50)	59
	P53	692.52716(50)	18	R13	911.00160(50)	-16
	P52	696.76022(50)	12	R14	913.01435(50)	-15
	P51	700.97172(50)	6	R15	914.98392(50)	-16
	P50	705.16135(50)	11	R16	916.91013(50)	-15
	P49	709.32807(50)	-39	R18	920.63150(50)	-13
	P48	713.47301(50)	10	R19	922.42624(50)	-10
	P47	717.59424(50)	4	R20	924.17663(50)	-16
	P46	721.69209(50)	16	R21	925.88263(50)	-12
	P45	725.76575(50)	3	R22	927.54388(50)	-13
	P44	729.81524(50)	6	R23	929.16031(50)	-7
	P43	733.83991(50)	-1	R24	930.73159(50)	-6
	P42	737.83964(50)	8	R25	932.25754(50)	-8
	P41	741.81377(50)	5	R26	933.73800(50)	-9
	P40	745.76208(50)	7	R27	935.17279(50)	-8
	P39	749.68415(50)	7	R28	936.56172(50)	-5
P38	753.57956(50)	3	R29	937.90461(50)	1	
P37	757.44812(50)	13	R30	939.20117(50)	-2	
P36	761.28913(50)	2	R31	940.45135(50)	-1	
P35	765.10256(50)	5	R32	941.65492(50)	0	
P34	768.88784(50)	2	R33	942.81174(50)	2	
P33	772.64482(50)	14	R34	943.92159(50)	0	
P32	776.37275(50)	1	R35	944.98457(50)	21	
P31	780.07159(50)	-4	R36	945.99995(50)	8	
P29	787.38047(50)	-3	R37	946.96804(50)	7	
P28	790.98979(50)	1	R38	947.88866(50)	15	
P27	794.56839(50)	-9	R39	948.76142(50)	8	
P26	798.11622(50)	-4	R40	949.58641(50)	9	
P25	801.63268(50)	-9	R41	950.36345(50)	14	
P24	805.11760(50)	-8	R42	951.09230(50)	12	
P23	808.57063(50)	-2	R43	951.77288(50)	9	
P22	811.99123(50)	-10	R44	952.40513(50)	11	
P21	815.37936(50)	-4	R45	952.98891(50)	15	
P20	818.73443(50)	-9	R46	953.52413(50)	26	
P19	822.05627(50)	-10	R47	954.01031(50)	5	
P18	825.34455(50)	-6	R48	954.44802(50)	21	
P17	828.59880(50)	-12	R49	954.83642(50)	1	
P16	831.81897(50)	-1	R50	955.17610(50)	12	
P15	835.00440(50)	-8	R51	955.46643(50)	2	
P13	841.27036(50)	-12	R52	955.70813(100)	52	
P12	844.35032(50)	-5	R53	955.89976(50)	26	
P11	847.39426(50)	-17	R55	956.13526(50)	22	
P10	850.40221(50)	-15	R58	956.11660(50)	-6	
P9	853.37365(50)	-21	R60	955.85520(100)	-71	
P8	856.30845(50)	-17	R61	955.65116(50)	32	
P7	859.20616(50)	-18	R62	955.39582(50)	-8	
P6	862.06649(50)	-24	R63	955.09177(50)	69	
P5	864.88929(50)	-20	P63	637.08050(50)	47	

$v = 3 \rightarrow 2$

TABLE 1 (continued)

J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}	J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}
P62	641.43448(50)	38	R4	875.21320(50)	-9
P61	645.77142(50)	23	R5	877.56817(50)	27
P60	650.09061(50)	-26	R6	879.88235(50)	15
P59	654.39259(50)	-14	R7	882.15584(50)	-10
P58	658.67705(50)	71	R8	884.38874(50)	-11
P57	662.94016(100)	-114	R9	886.58060(50)	-10
P56	667.18712(50)	-7	R10	888.73113(50)	-10
P55	671.41350(50)	-10	R11	890.84013(50)	-8
P54	675.62027(50)	14	R12	892.90734(50)	-6
P53	679.80640(50)	4	R13	894.93247(50)	-8
P52	683.97197(50)	6	R14	896.91535(50)	-9
P51	688.11522(100)	-114	R15	898.85576(50)	-7
P50	692.23944(50)	12	R16	900.75346(50)	-5
P49	696.34045(50)	7	R17	902.60817(50)	-6
P48	700.41927(50)	10	R18	904.41976(50)	-4
P47	704.47535(50)	7	R19	906.18792(50)	-6
P46	708.50834(50)	1	R20	907.91254(50)	-3
P45	712.51815(50)	23	R21	909.59335(50)	0
P44	716.50367(50)	-1	R22	911.23010(50)	-2
P43	720.46523(50)	1	R23	912.82266(50)	-2
P42	724.40076(150)	-140	R24	914.37083(50)	1
P41	728.31423(50)	11	R25	915.87433(50)	-2
P40	732.20075(50)	2	R26	917.33306(50)	-1
P39	736.06160(50)	-1	R27	918.74679(50)	-2
P38	739.89648(50)	9	R28	920.11539(50)	3
P37	743.70475(50)	4	R29	921.43861(50)	5
P36	747.48629(50)	11	R30	922.71629(50)	7
P35	751.24050(50)	4	R31	923.94823(50)	7
P34	754.96726(50)	8	R32	925.13427(50)	5
P33	758.66554(50)	-44	R33	926.27433(50)	10
P32	762.33658(50)	9	R34	927.36815(50)	12
P31	765.97844(50)	7	R35	928.41551(50)	5
P30	769.59128(50)	1	R36	929.41649(50)	12
P29	773.17489(50)	7	R37	930.37073(50)	14
P28	776.72871(50)	3	R38	931.27823(50)	24
P27	780.25263(50)	11	R39	932.13854(50)	12
P25	787.20868(50)	-3	R40	932.95179(50)	4
P24	790.64043(50)	4	R41	933.71793(50)	9
P23	794.04065(50)	-2	R42	934.43673(50)	18
P22	797.40920(50)	-3	R43	935.10785(50)	8
P21	800.74569(50)	-3	R44	935.73142(50)	6
P20	804.04978(50)	-4	R45	936.30716(50)	-6
P19	807.32117(50)	-4	R46	936.83529(50)	6
P18	810.55952(50)	-4	R47	937.31524(50)	-4
P17	813.76451(50)	-3	R48	937.74720(50)	-7
P16	816.93581(50)	-3	R49	938.13106(50)	-3
P15	820.07305(50)	-8	R50	938.46692(50)	26
P14	823.17607(50)	-5	R51	938.75398(50)	11
P13	826.24442(50)	-5	R54	939.32502(50)	44
P12	829.27779(50)	-10	R55	939.41816(100)	58
P11	832.27601(50)	-5	R58	939.40280(100)	-116
P10	835.23860(50)	-9	P61	633.87765(50)	-1
P8	841.05602(50)	-7	P60	638.12641(100)	-83
P7	843.91016(50)	-11	P59	642.35907(50)	-35
P6	846.72833(100)	62	P58	646.57356(50)	-25
P5	849.50808(50)	-4	P57	650.76959(50)	-38
P4	852.25109(50)	-11	P56	654.94728(50)	-24
P3	854.95662(50)	-6	P55	659.10622(50)	19
P2	857.62348(100)	-79	P54	663.24355(150)	-156
P1	860.25368(50)	-1	P53	667.36412(50)	-24
R0	865.39691(50)	-1	P52	671.46338(50)	1
R1	867.90996(50)	-21	P51	675.54147(50)	-27
R2	870.38398(50)	-18	P50	679.59883(50)	-27
R3	872.81853(50)	-10	P49	683.63493(50)	-10

 $\nu = 4 \rightarrow 3$

TABLE 1 (continued)

J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹	J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹
P48	687.64923(50)	8	R18	888.53957(50)	4
P47	691.64100(50)	-7	R19	890.28173(50)	6
P46	695.61031(50)	-9	R20	891.98095(50)	6
P45	699.55677(50)	1	R21	893.63709(50)	11
P44	703.47971(50)	-7	R22	895.24986(50)	13
P43	707.37895(50)	-11	R23	896.81903(50)	9
P42	711.25435(50)	12	R24	898.34452(50)	11
P41	715.10502(50)	10	R25	899.82602(50)	6
P40	718.93082(50)	7	R26	901.26345(50)	6
P39	722.73143(50)	8	R27	902.65654(50)	3
P38	726.50642(50)	7	R28	904.00516(50)	0
P37	730.25548(50)	8	R29	905.30926(50)	12
P36	733.97812(50)	1	R30	906.56825(50)	-3
P35	737.67426(50)	13	R31	907.78243(50)	2
P34	741.34315(50)	4	R33	910.07504(50)	5
P33	744.98473(50)	5	R34	911.15319(50)	8
P32	748.59889(50)	40	R35	912.18567(50)	9
P31	752.18426(50)	7	R36	913.17222(50)	-2
P30	755.74152(50)	10	R37	914.11269(50)	-26
P28	762.76918(50)	8	R38	915.00775(50)	19
P27	766.23995(100)	110	R39	915.85593(50)	-1
P26	769.67877(50)	0	R40	916.65813(50)	19
P25	773.08856(50)	6	R41	917.41343(50)	-2
P24	776.46786(50)	14	R42	918.12241(50)	9
P23	779.81612(50)	4	R44	919.40004(50)	34
P22	783.13355(50)	29	R45	919.96811(50)	14
P21	786.41897(50)	4	R46	920.48895(50)	-20
P20	789.67280(50)	4	R47	920.96316(50)	3
P19	792.89451(50)	9	R48	921.39014(50)	34
P18	796.08353(50)	-6	R49	921.76922(50)	13
P17	799.23978(50)	-18	R50	922.10129(50)	41
P16	802.36319(50)	-2	R52	922.62144(50)	-22
P15	805.45303(50)	1	R54	922.95179(50)	29
P14	808.50912(50)	4	P57	638.86621(50)	31
P13	811.53103(50)	-5	P56	642.97601(50)	-20
P12	814.51865(50)	-6	P55	647.06761(50)	-31
P11	817.47133(50)	-35	P54	651.14147(100)	84
P10	820.38974(50)	7	P53	655.19459(50)	64
P9	823.27215(50)	-24	P52	659.22747(50)	-2
P8	826.11958(50)	4	P51	663.24355(300)	271
P7	828.93075(50)	-8	P50	667.23339(50)	-22
P6	831.70600(50)	3	P49	671.20542(50)	-1
P5	834.44486(50)	19	P48	675.15624(50)	35
P4	837.14646(50)	-17	P47	679.08464(50)	3
P3	839.81170(50)	11	P46	682.99066(100)	-56
P2	842.43925(50)	0	P45	686.87564(50)	31
P1	845.02984(50)	49	P44	690.73652(50)	-3
R0	850.09554(50)	-19	P43	694.57445(50)	-8
R1	852.57155(50)	7	P42	698.38898(50)	11
R2	855.00854(50)	-3	P41	702.17991(100)	70
R3	857.40662(50)	-14	P40	705.94515(50)	-3
R4	859.76600(50)	24	P39	709.68645(50)	4
R5	862.08542(50)	9	P38	713.40242(50)	-12
R6	864.36526(50)	5	P37	717.09308(50)	-12
R7	866.60519(50)	4	P36	720.75821(50)	19
R8	868.80486(50)	-5	P34	728.00889(50)	13
R9	870.96427(50)	5	P33	731.59387(50)	-8
R10	873.08284(50)	-2	P32	735.15166(50)	-24
R11	875.16049(50)	-10	P31	738.68227(50)	3
R12	877.19718(50)	2	P30	742.18469(50)	6
R13	879.19241(50)	7	P29	745.65885(50)	12
R15	883.05771(50)	8	P28	749.10425(50)	6
R16	884.92738(50)	9	P27	752.52062(50)	-5
R17	886.75478(50)	12	P26	755.90785(50)	1

$\nu = 5 \rightarrow 4$

TABLE 1 (continued)

J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}	J''	Obs./ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}
P24	762.59288(50)	1	R45	903.96352(50)	16
P23	765.89020(50)	12	R46	903.47840(50)	45
P22	769.15714(50)	5	R47	904.94617(50)	9
P21	772.39227(50)	4	R48	905.36761(50)	-6
P20	775.59618(50)	-34	R49	905.74250(50)	-12
P29	778.76916(50)	-3	R50	906.07049(50)	-35
P18	781.90999(50)	6	R51	906.35198(50)	-27
P17	785.01842(50)	2	R55	907.00830(50)	7
P16	788.09442(50)	11	P56	631.26624(50)	0
P15	791.13745(50)	11	P54	639.29894(100)	-76
P14	794.14719(50)	2	P52	647.25691(50)	-39
P13	797.12355(50)	4	P51	651.20753(100)	85
P12	800.06611(50)	6	P50	655.13622(50)	30
P11	802.97466(50)	18	P49	659.04535(100)	71
P10	805.84856(50)	5	P46	670.64369(50)	-18
P9	808.68758(50)	-26	P45	674.46635(50)	-35
P8	811.49219(50)	2	P44	678.26686(50)	-25
P6	816.99486(50)	15	P43	682.04524(50)	51
P5	819.69328(100)	95	P42	685.79951(50)	32
P4	822.35384(50)	3	P41	689.53022(50)	9
P3	824.98004(150)	118	P40	693.23728(50)	12
P2	827.56823(100)	101	P39	696.91967(50)	-27
R1	837.54801(50)	-17	P38	700.57774(50)	-35
R2	839.94955(50)	58	P37	704.21097(50)	-29
R3	842.31126(50)	-19	P36	707.81909(50)	1
R4	844.63543(50)	6	P35	711.40128(50)	7
R5	846.92047(50)	0	P34	714.95743(50)	14
R6	849.16668(50)	19	P33	718.48678(50)	-18
R7	851.37325(50)	5	P32	721.99000(50)	11
R8	853.54035(50)	1	P31	725.46568(50)	-3
R9	855.66776(50)	9	P30	728.91460(50)	51
R11	859.80190(50)	-5	P29	732.33460(50)	-9
R12	861.80857(50)	14	P28	735.72725(50)	9
R13	863.77432(50)	15	P27	739.09118(50)	1
R14	865.69880(50)	-13	P26	742.42629(50)	-8
R15	867.58258(50)	10	P25	745.73224(50)	-21
R16	869.42471(50)	9	P23	752.25605(50)	17
R17	871.22522(50)	9	P22	755.47261(50)	3
R18	872.98382(50)	4	P20	761.81424(50)	-10
R19	874.70051(50)	14	P19	764.93689(200)	-187
R20	876.37479(50)	9	P18	768.03182(50)	5
R21	878.00655(50)	0	P17	771.09320(50)	12
R22	879.59579(50)	6	P16	774.12229(50)	-7
R24	882.64532(50)	3	P15	777.11933(50)	2
R25	884.10528(50)	-1	P13	783.01522(50)	24
R26	885.52187(50)	2	P12	785.91301(50)	-9
R27	886.89474(50)	-5	P11	788.77891(150)	124
R28	888.22390(50)	-3	P10	791.60833(50)	-8
R29	889.50909(50)	-1	P9	794.40495(50)	-6
R30	890.75008(50)	-4	P8	797.16733(50)	15
R31	891.94677(50)	-6	P7	799.89493(50)	28
R32	893.09916(50)	10	P6	802.58730(50)	19
R33	894.20679(50)	13	P5	805.24087(350)	-342
R34	895.26944(50)	-2	P4	807.86617(50)	26
R35	896.28763(50)	31	P2	813.00221(100)	87
R36	897.26012(50)	3	R2	825.19847(50)	-3
R37	898.18763(50)	1	R3	827.52599(50)	14
R38	899.07002(50)	25	R4	829.81527(50)	2
R39	899.90634(50)	-7	R5	832.06651(50)	8
R40	900.69714(50)	-26	R6	834.27886(50)	-30
R41	901.44250(50)	-12	R8	838.58819(50)	-5
R42	902.14189(50)	-5	R9	840.68389(50)	-23
R43	902.79536(50)	11	R10	842.74058(50)	1
R44	903.40252(50)	9	R11	844.75707(50)	-30

 $\nu = 6 \rightarrow 5$

TABLE 1 (continued)

J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹	J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹
R12	846.72833(600)	-594	P21	745.21175(50)	-23
R13	848.67116(50)	10	P20	748.31931(50)	-12
R14	850.56750(50)	0	P19	751.39546(100)	-86
R15	852.42330(50)	-9	P18	754.44228(50)	-7
R16	854.23959(50)	10	P17	757.44812(1000)	-908
R17	856.01250(50)	-11	P16	760.44010(50)	-47
R18	857.75264(700)	712	P15	763.39169(50)	-45
R19	859.43674(50)	-27	P14	766.31080(100)	-82
R20	861.08708(50)	18	P13	769.19877(50)	7
R21	862.69487(50)	-9	P12	772.05255(50)	-53
R22	864.26104(50)	2	P11	774.87448(50)	1
R23	865.78496(50)	10	P10	777.66275(50)	17
R24	867.26622(50)	-9	P9	780.41688(50)	-23
R25	868.70521(50)	3	P8	783.13355(500)	-423
R26	870.10125(50)	-3	P7	785.82366(100)	-63
R27	871.45421(50)	-22	R0	806.05185(200)	-178
R28	872.76452(50)	6	R3	813.04233(100)	-76
R29	874.03130(50)	10	R6	819.69328(300)	-302
R30	875.25442(50)	-6	R7	821.83811(50)	-6
R31	876.43369(50)	-44	R9	826.00653(50)	-11
R32	877.56817(200)	-183	R10	828.03300(50)	22
R33	878.66191(50)	-1	R11	830.02000(50)	13
R34	879.70976(50)	2	R12	831.96699(100)	-71
R35	880.71316(50)	-16	R13	833.87574(50)	-29
R36	881.67234(50)	-17	R14	835.74396(50)	-69
R37	882.58714(50)	-3	R15	837.57308(50)	-26
R38	883.45736(50)	20	R16	839.36132(100)	-56
R39	884.28195(50)	-40	R17	841.10989(50)	-18
R40	885.06264(50)	4	R18	842.81765(50)	-3
R41	885.79740(50)	-41	R19	844.48452(50)	-1
R42	886.48648(150)	-136	R20	846.11019(50)	-22
R44	887.73275(100)	85	R21	847.69396(150)	-116
R45	888.28561(50)	-12	R22	849.23796(50)	-51
R47	889.25732(100)	89	R23	850.74056(50)	29
R50	890.36885(50)	14	R24	852.20033(50)	1
R51	890.64793(50)	48	R25	853.61839(50)	-6
R54	891.20663(50)	8	R26	854.99312(150)	-136
P49	647.14517(100)	-58	R27	856.33010(200)	188
P48	650.97177(50)	-19	R28	857.62348(400)	396
P47	654.77918(200)	187	R29	858.86836(50)	17
P46	658.56125(50)	-20	R30	860.07412(50)	5
P45	662.32515(100)	116	R31	861.23685(50)	-16
P44	666.06510(100)	54	R33	863.43413(100)	72
P42	673.47846(50)	12	R34	864.46643(50)	-14
P41	677.15168(100)	88	R35	865.45592(50)	-25
P40	680.79849(150)	-134	R36	866.40256(50)	49
P39	684.42469(50)	-38	R37	867.30403(50)	-11
P38	688.02566(50)	-50	R38	868.16397(200)	173
P37	691.60208(100)	-66	R39	868.97640(50)	16
P36	695.15490(50)	45	R40	869.74394(250)	-208
P35	698.68078(50)	-17	R41	870.47184(50)	40
P34	702.17991(200)	-197	R42	871.15340(100)	100
P33	705.65682(50)	-8	R43	871.78846(50)	-32
P32	709.10513(50)	-52	R44	872.38031(50)	-17
P31	712.51815(1000)	-965	R46	873.42902(50)	-37
P30	715.92281(50)	-19	R49	874.66493(50)	-19
P29	719.29063(50)	-29	P42	661.42085(150)	141
P28	722.63170(50)	49	P38	675.74003(50)	10
P27	725.94266(100)	-89	P35	686.23006(100)	98
P26	729.23121(350)	362	P33	693.09693(50)	-3
P25	732.48296(50)	-5	P32	696.49270(50)	31
P24	735.70932(50)	-17	P30	703.20429(50)	-28
P23	738.90654(50)	-15	P28	709.80945(50)	-11
P22	742.07481(50)	52	P27	713.07091(50)	-12

$\nu = 7 \rightarrow 6$

$\nu = 8 \rightarrow 7$

TABLE 1 (continued)

J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹	J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹
P26	716.30457(50)	-14	P15	896.44556(50)	26
P25	719.51041(50)	14	P14	900.06386(50)	8
P24	722.68777(50)	38	P13	903.64038(50)	27
P23	725.83593(50)	19	P12	907.17407(50)	17
P22	728.95561(50)	61	P11	910.66485(50)	11
P21	732.04517(50)	30	P9	917.51612(50)	8
P20	735.10512(50)	11	P8	920.87587(50)	14
P19	738.13524(50)	12	P6	927.46149(50)	22
P18	741.13479(50)	-10	P5	930.68665(50)	29
P17	744.10391(50)	-10	P4	933.86603(50)	18
P15	749.95028(150)	121	P3	936.99984(50)	47
P14	752.82567(150)	126	P2	940.08696(50)	42
R3	798.85770(150)	142	R1	951.96648(200)	178
R4	801.07990(150)	160	R2	954.81511(50)	26
R8	809.60458(1000)	1080	R3	957.61714(50)	58
R9	811.62911(100)	81	R4	960.36968(50)	19
R10	813.62541(100)	79	R5	963.07337(50)	7
R13	819.37905(300)	-305	R7	968.33239(50)	14
R14	821.22278(100)	-59	R8	970.88707(50)	32
R15	823.02561(50)	29	R9	973.39100(50)	16
R19	829.83516(100)	-69	R10	975.84433(50)	13
R25	838.83764(50)	-28	R11	978.24675(50)	23
R27	841.50984(100)	91	R12	980.59770(50)	20
R28	842.78178(50)	-6	R13	982.89712(50)	29
R29	844.01424(150)	146	R14	985.14439(50)	17
R30	845.20204(50)	44	R15	987.33946(50)	9
R31	846.34856(50)	42	R16	989.48225(50)	25
R32	847.45203(50)	-21	R17	991.57200(50)	18
R34	849.53230(50)	-24	R18	993.60879(50)	25
R37	852.33205(100)	97	R19	995.59199(50)	9
R38	853.17689(100)	-64	R20	997.52168(50)	7
			R21	999.39761(50)	19
			R22	1001.21912(50)	6
			R23	1002.98612(50)	-15
			R24	1004.69906(50)	27
			R25	1006.35654(50)	15
			R26	1007.95899(50)	18
			R27	1009.50605(50)	24
			R28	1010.99719(50)	2
			R29	1012.43284(50)	19
			R30	1013.81195(50)	-7
			R31	1015.13504(50)	-3
			R32	1016.40130(50)	-28
			R33	1017.61128(50)	-7
			R34	1018.76427(50)	11
			R35	1019.85968(50)	-14
			R36	1020.89828(50)	14
			R37	1021.87841(50)	-52
			R38	1022.80178(50)	-22
			R39	1023.66705(50)	-13
			R40	1024.47393(50)	-36
			R41	1025.22300(50)	-17
			R42	1025.91354(50)	-12
			R43	1026.54473(100)	-88
			R44	1027.11936(100)	50
			R45	1027.63336(50)	9
			R46	1028.08907(50)	36
			R47	1028.48432(50)	-72
			R48	1028.82102(100)	-113
			R49	1029.09819(150)	-171
			R50	1029.31757(50)	-62
			R51	1029.47703(50)	11
			R52	1029.57530(50)	-67
			P45	755.43538(100)	-67

⁶LiF
 $\nu = 1 \rightarrow 0$ $\nu = 2 \rightarrow 1$

TABLE I (continued)

J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹	J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹
P43	764.61478(50)	-45	R28	992.16338(50)	-8
P41	773.67949(50)	-60	R29	993.57652(50)	-9
P40	778.16834(50)	-10	R30	994.93450(50)	-6
P39	782.62662(50)	-13	R31	996.23725(50)	15
P38	787.05450(50)	-6	R32	997.48431(50)	29
P37	791.45148(50)	9	R33	998.67506(50)	-5
P36	795.81657(50)	-19	R34	999.81007(50)	-11
P35	800.14984(50)	-37	R35	1000.88915(50)	12
P34	804.45136(50)	8	R36	1001.91146(50)	-1
P33	808.71805(150)	-145	R37	1002.87728(50)	-5
P32	812.95455(50)	13	R38	1003.78611(50)	-30
P31	817.15543(50)	-15	R39	1004.63850(50)	-5
P30	821.32216(50)	-37	R40	1005.43360(50)	1
P29	825.45465(50)	-16	R41	1006.17163(50)	27
P28	829.55201(50)	3	R42	1006.85155(50)	-16
P27	833.61364(50)	3	R43	1007.47425(50)	-24
P26	837.63928(50)	4	R44	1008.03932(50)	-23
P25	841.62865(50)	20	R45	1008.54547(150)	-129
P24	845.58079(50)	0	R46	1008.99703(150)	105
P23	849.49399(200)	-185	R47	1009.38662(50)	-47
P21	857.21231(50)	-5	R48	1009.71917(50)	-80
P20	861.01304(50)	5	P41	758.75890(100)	99
P19	864.77476(50)	13	P40	763.17149(50)	-22
P18	868.49715(50)	28	P38	771.91079(50)	14
P17	872.17929(50)	-2	P37	776.23471(50)	-13
P16	875.82136(50)	-17	P36	780.52789(50)	-33
P15	879.42319(50)	6	P34	789.02044(50)	-25
P14	882.98366(50)	-4	P33	793.21835(50)	-52
P13	886.50302(50)	16	P32	797.38225(250)	-214
P12	889.98011(50)	-9	P31	801.51430(250)	-252
P11	893.41533(50)	0	P30	805.61544(50)	-25
P9	900.15735(50)	-7	P29	809.68058(50)	1
P8	903.46399(50)	37	P28	813.71070(50)	-31
P7	906.72608(50)	0	P27	817.70707(50)	50
P6	909.94425(50)	-18	P26	821.66970(300)	289
P5	913.11846(50)	17	P25	825.59133(50)	2
P4	916.24665(100)	-66	P24	829.47955(50)	-7
P3	919.33092(50)	-18	P22	837.14646(50)	45
R1	934.05897(50)	-45	P21	840.92343(50)	19
R2	936.86533(100)	83	P19	848.36390(50)	22
R3	939.62222(50)	29	P18	852.02613(50)	7
R4	942.33100(50)	-36	P17	855.64945(50)	11
R6	947.60494(50)	-1	P16	859.23324(50)	13
R7	950.16843(50)	-2	P15	862.77709(50)	11
R8	952.68273(50)	7	P14	866.28072(50)	19
R9	955.14718(50)	-9	P13	869.74394(50)	55
R10	957.56191(50)	-6	P12	873.16500(50)	-15
R11	959.92643(50)	-2	P11	876.54562(50)	19
R12	962.24050(50)	8	P9	883.18000(50)	-2
R13	964.50353(50)	-3	P8	886.43358(50)	1
R14	966.71581(50)	21	P7	889.64414(50)	3
R15	968.87667(50)	44	P6	892.81130(50)	1
R16	970.98505(50)	-13	P5	895.93473(50)	0
R17	973.04232(50)	15	P4	899.01544(150)	136
R18	975.04697(50)	6	P2	905.04087(200)	184
R19	976.99921(50)	8	R3	922.01884(50)	5
R20	978.89871(50)	13	R4	924.68524(50)	-24
R21	980.84527(50)	29	R5	927.30477(50)	14
R22	982.53819(50)	11	R7	932.39898(50)	-8
R23	984.27761(50)	-2	R8	934.87365(50)	-5
R24	985.96322(50)	-14	R9	937.29833(150)	-121
R25	987.59499(50)	-6	R10	939.67647(50)	19
R26	989.17244(50)	-2	R11	942.00368(50)	7
R27	990.69572(50)	39	R12	944.28135(50)	11

$v = 3 \rightarrow 2$

TABLE 1 (concluded)

J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹	J''	Obs./cm ⁻¹	Obs. - calc. /10 ⁻⁵ cm ⁻¹
R13	946.50882(50)	-5	R28	973.73717(50)	-32
R14	948.68641(50)	19	R29	975.12887(50)	2
R15	950.81344(50)	45	R30	976.46613(50)	22
R16	952.88892(50)	0	R31	977.74904(100)	58
R17	954.91365(50)	-6	R32	978.97634(50)	4
R18	956.88709(50)	-2	R33	980.14903(50)	-19
R19	958.80858(50)	-26	R34	981.26726(50)	22
R20	960.67877(50)	14	R35	982.32871(100)	-85
R21	962.49611(50)	-13	R36	983.33625(50)	-34
R22	964.26116(50)	-24	R38	985.18360(50)	9
R23	965.97345(50)	-43	R39	986.02215(100)	-91
R24	967.63365(50)	23	R40	986.80581(50)	-63
R25	969.23974(50)	-4	R41	987.53253(100)	-98
R26	970.79251(50)	-22	R43	988.81787(50)	-23
R27	972.29165(50)	-39			

^aNumbers in parentheses are estimated uncertainties.

sities of the emission lines increased with higher temperature. The maximum temperature which is accessible with our current system is about 1500 K and it was at this temperature that the best emission spectrum was obtained. All emission signals decreased rapidly as the furnace cooled and disappeared at about 1200 K.

III. Results

The emission features were observed in the region where the infrared spectrum of MgF₂ was expected (Figs. 1, 2). However, the high resolution spectrum showed a pattern typical of a diatomic molecule. After some thought and a literature search we found that the emission spectrum belonged to LiF (11). The presence of Li in our system was puzzling but in a previous experiment the alumina tube was used to vaporize LiNC. Despite careful cleaning of the cell, some Li remained on the walls of the alumina tube. At higher temperatures the MgF₂ could react with water (which is always present in the system) and form HF. The HF can react with the Li impurities and produce LiF. Direct reduction of MgF₂ with Li is also possible but MgF was not detected. An additional reaction of HF with the alumina tube took place to form AlF. The emission of AlF was observed in the spectrum (17).

The spectral analysis program PC-DECOMP, developed by J. W. Brault, was used for data analysis. The rotational line profiles were fit to Voigt lineshape functions. The strong lines show a "ringing" caused by the (sin x/x) lineshape function of the Fourier transform spectrometer. The ringing was eliminated by using the "filter fitting" routine available in PC-DECOMP. The signal-to-noise ratio for the strong lines was better than 1000 and the resulting resolution-enhanced line width was 0.0005 cm⁻¹. The HF lines were used for absolute calibration (+/-0.0002 cm⁻¹) of the spectrum. For this calibration the HF absorption lines, taken in a spectrum at lower temperature, were calibrated against CO₂ (18).

The assignment of the vibrational bands of the main isotopomer ⁷LiF was straightforward (Table 1). Data reduction was made by using Dunham's equation (19)

$$[1] \quad T(\nu, J) = \sum_{i,j} Y_{ij} \left(\nu + \frac{1}{2} \right)^i [J(J+1)]^j$$

Pure rotational transitions reported by Pearson and Gordy (20)

TABLE 2. Dunham constants for LiF (in cm⁻¹)^a

Constant	Value	
	⁷ LiF	⁶ LiF
Y_{10}	910.57272(10)	964.30621(22)
Y_{20}	-8.207956(46)	-9.19819(12)
Y_{30}	0.569166(82)	0.64808(20)
Y_{40}	-0.28437(49)E-3	
Y_{01}	1.34525715(57)	1.5087385(14)
Y_{11}	-0.02028749(19)	-0.02409773(41)
Y_{21}	0.156247(64)E-3	0.19766(15)E-3
Y_{31}	-0.3860(88)E-6	-0.736(21)E-6
Y_{41}	-0.870(45)E-8	
Y_{02}	-0.1174389(33)E-4	-0.147719(13)E-4
Y_{12}	0.113878(98)E-6	0.15246(29)E-6
Y_{22}	0.720(23)E-9	0.619(33)E-9
Y_{32}	-0.306(13)E-10	
Y_{03}	0.30449(55)E-10	0.4300(35)E-10
Y_{13}	0.836(15)E-12	0.1120(70)E-11
Y_{23}	-0.270(27)E-13	

^aEach isotopomer was fit independently. One standard deviation in parentheses.

were included in the fit. The Dunham coefficients obtained from this fit are shown in Table 2.

The assignment of the vibrational bands of the minor isotopomer, ⁶LiF, was carried out by comparing predicted line positions with the unassigned lines in the spectrum. The line positions for ⁶LiF are also shown in Table 1. The mass-dependent Dunham coefficients Y_{ij} are listed in Table 2.

Finally, all infrared and microwave data for LiF was combined and fit to the mass-reduced Dunham expression including Watson's correction due to the breakdown of the Born-Oppenheimer approximation (21, 22)

$$[2] \quad Y_{ij} = \mu^{[(i+2j)/2]} U_{ij} \left[1 + \left(\frac{m_e}{m_{Li}} \right) \Delta_{ij} \right]$$

where μ is the reduced mass, U_{ij} are the mass-independent parameters, m_{Li} is the atomic mass of Li (23), m_e is the electron mass, and Δ_{ij} is the mass scaling factor for the Li atom. The

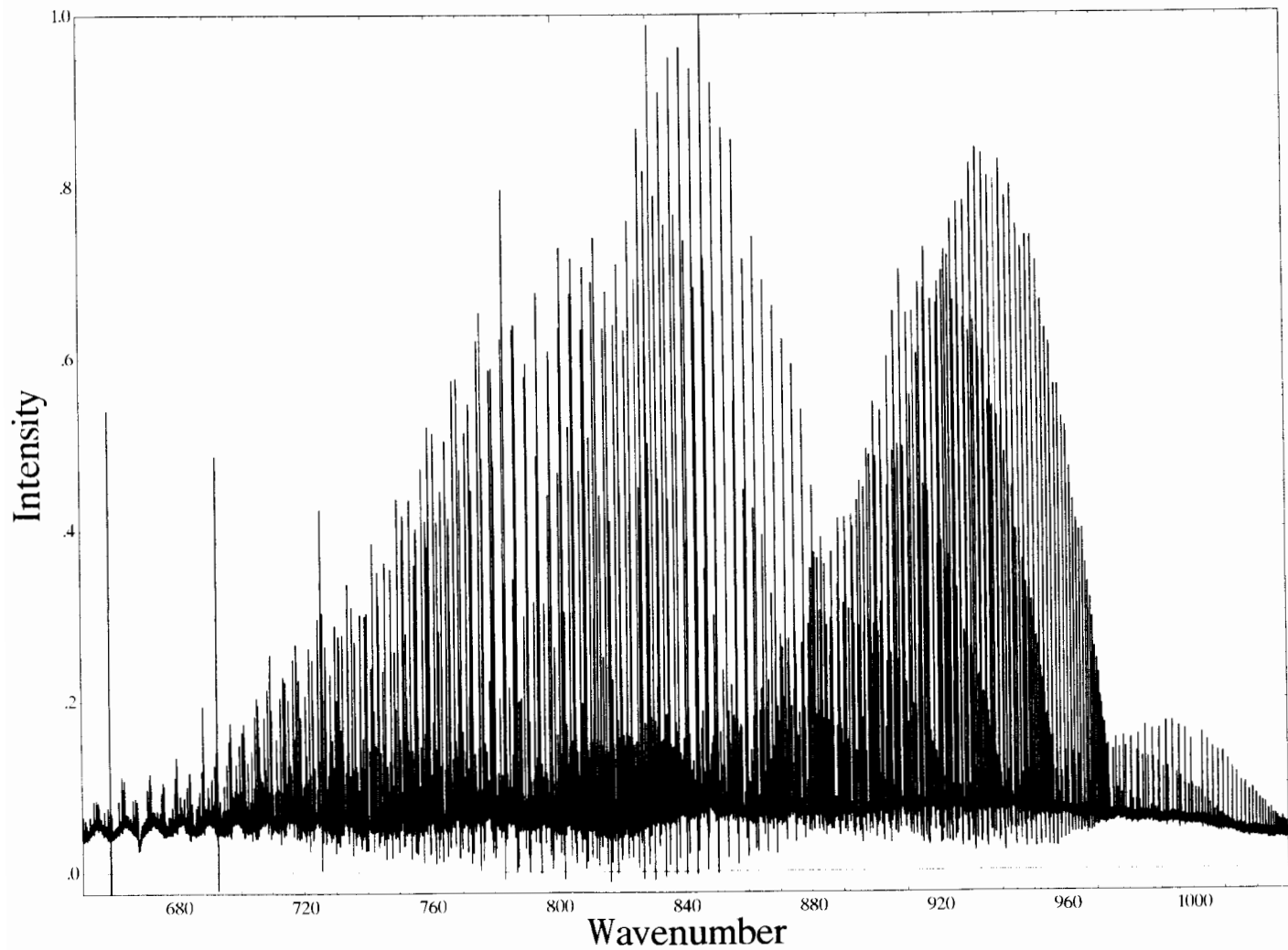


FIG. 1. Infrared spectrum of ${}^7\text{LiF}$.

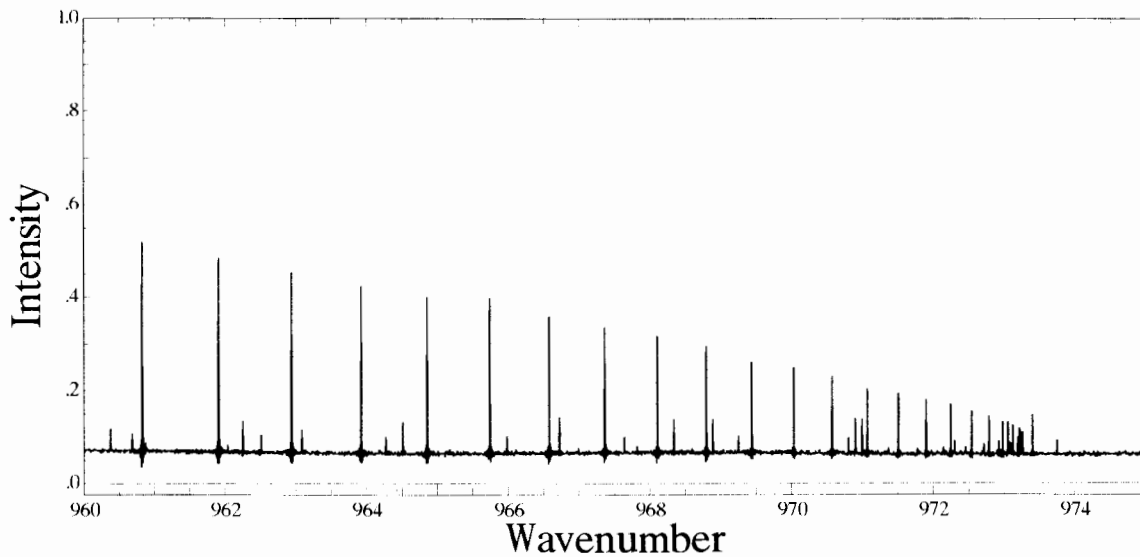


FIG. 2. Infrared spectrum of ${}^6\text{LiF}$.

TABLE 3. Mass-reduced Dunham constants for LiF (in cm^{-1})^a

Constant	Our work
U_{10}	2061.15184(33)
U_{20}	-42.05641(19)
U_{30}	0.660241(80)
U_{40}	-7.480(11)E-3
U_{01}	6.8926542(36)
U_{11}	-0.2352993(16)
U_{21}	4.1016(14)E-3
U_{31}	-0.2259(44)E-4
U_{41}	-0.1199(52)E-5
U_{02}	-0.3083223(30)E-3
U_{12}	0.67650(42)E-5
U_{22}	0.991(25)E-7
U_{32}	-0.995(32)E-8
U_{03}	0.40964(30)E-8
U_{13}	0.2566(34)E-9
U_{23}	-0.186(17)E-10
Δ_{10}	0.0417(17)
Δ_{01}	0.3595(58)

^aOne standard deviation in parentheses.

TABLE 4. Pure rotational transitions of HF

J'	J''	Observed/ cm^{-1}	Obs. - calc. / 10^{-5}cm^{-1}	Uncertainty (1σ) / 10^{-5}cm^{-1}
1	0	41.1109832 ^a	0.181	0.15
2	1	82.1711179 ^a	0.139	0.3
3	2	123.12967 ^a	-1	0.5
4	3	163.93616 ^a	0	0.5
5	4	204.54045 ^b	1	10
6	5	244.89283 ^b	0	10
7	6	284.94444 ^b	23	15
13	12	516.28061 ^c	-14	20
14	13	552.92046 ^c	9	20
15	14	588.89904 ^c	-1	20
16	15	624.17733 ^c	11	20
17	16	658.71682 ^c	4	20
18	17	692.48121 ^c	8	20
19	18	725.43557 ^c	38	20
20	19	757.54540 ^c	-13	20
21	20	788.78013 ^c	-15	20
22	21	819.10918 ^c	-8	20
23	22	848.50386 ^c	-8	20
24	23	876.93745 ^c	-5	20
25	24	904.38481 ^c	5	20
27	26	956.22816 ^d	-3	13
28	27	980.5832 ^e	74	75
29	28	1003.8672 ^e	62	50
31	30	1047.1570 ^e	-169	125
32	31	1067.1377 ^e	-11	50
33	32	1085.98903 ^d	0	13

^aTunable far infrared spectroscopy (ref. 12).^bFourier transform spectroscopy (ref. 12).^cFourier transform spectroscopy (our work).^dDiode laser heterodyne spectroscopy (ref. 14).^eDiode laser spectroscopy (ref. 12).TABLE 5. Rotational constants of ground state HF^a

Constant	Our work / cm^{-1}	Jennings <i>et al.</i> (12) / cm^{-1}
B_0	20.5597298(8)	20.5597300(25)
D_0	2.119901(36)E-3	2.11991(14)E-3
H_0	0.163483(116)E-6	0.16334(50)E-6
L_0	-0.14974(132)E-10	-0.1473(66)E-10
M_0	0.1052(50)E-14	0.94(27)E-15

^aTwo standard deviations in parentheses.

data required two Δ 's, one for U_{10} (" ω_e ") and one for U_{01} (" B_e "). The final mass-independent coefficients and the Δ 's are listed in Table 3.

The energy levels of HF were fit to the expression (24)

$$[3] \quad E_v(J) = B_v J(J+1) - D_v [J(J+1)]^2 + H_v [J(J+1)]^3 + L_v [J(J+1)]^4 + M_v [J(J+1)]^5$$

The fit includes all lines listed in Table 4 and Table 5 shows the determined constants.

IV. Conclusion

The high-resolution infrared emission spectra of LiF and HF were observed with a Fourier transform instrument. The ro-vibrational transitions of the fundamental band and of seven hot bands for ⁷LiF and of three bands for ⁶LiF were rotationally analyzed. Pure rotational transitions for HF were also observed and enabled us to calculate an improved set of molecular constants. The accuracy of the line positions allows the use of HF as an absolute wavenumber standard.

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1. S. E. VEAZEY and W. GORDY. *Phys. Rev. A*, **138**, 1303 (1965).
2. A. J. HEBERT, F. J. LOVAS, C. A. MELENDRES, C. D. HOLLOWELL, T. L. STORY, JR., and K. STREET, JR. *J. Chem. Phys.* **48**, 2824 (1968).
3. G. L. VIDALE. *J. Phys. Chem.* **64**, 314 (1960).
4. L. WHARTON, W. KLEMPERER, L. P. GOLD, R. STRAUCH, J. J. GALLAGHER, and V. E. DERR. *J. Chem. Phys.* **38**, 1203 (1963).
5. M. J. LINEVSKY. *J. Chem. Phys.* **34**, 587 (1961).
6. M. J. LINEVSKY. *J. Chem. Phys.* **38**, 658 (1963).
7. A. SNELSON and K. S. PITZER. *J. Phys. Chem.* **67**, 882 (1963).
8. S. SCHLICK and O. SCHNEPP. *J. Chem. Phys.* **41**, 463 (1964).
9. A. SNELSON. *J. Chem. Phys.* **46**, 3652 (1967).
10. S. ABRAMOWITZ, N. ACQUISTA, and I. W. LEVIN. *J. Res. Nat. Bur. Stand. Sect. A*, **72**, 487 (1968).
11. A. MAKI. *J. Mol. Spectrosc.* **102**, 361 (1983).
12. D. A. JENNINGS, K. M. EVENSON, L. R. ZINK, C. DEMUYNCK, J. L. DESTOMBES, B. LEMOINE, and J. W. C. JOHNS. *J. Mol. Spectrosc.* **122**, 477 (1987).

13. U. K. SENGUPTA, P. K. DAS, and K. N. RAO. *J. Mol. Spectrosc.* **74**, 322 (1979).
14. D. A. JENNINGS and J. S. WELLS. *J. Mol. Spectrosc.* **130**, 267 (1988).
15. J. S. WELLS, F. R. PETERSON, A. G. MAKI, and D. J. SUKLE. *Appl. Opt.* **20**, 1676 (1981).
16. C. I. FRUM, R. ENGLEMAN, JR., and P. F. BERNATH. *J. Chem. Phys.* **93**, 5457 (1990).
17. H. G. HEDDERICH, C. I. FRUM, R. ENGLEMAN, JR., and P. F. BERNATH. To be published.
18. G. GUELACHVILI and K. N. RAO. *Handbook of infrared standards*. Academic Press, Orlando, FL. 1986.
19. J. L. DUNHAM. *Phys. Rev.* **41**, 721 (1932).
20. E. F. PEARSON and W. GORDY. *Phys. Rev.* **177**, 52 (1969).
21. J. K. G. WATSON. *J. Mol. Spectrosc.* **45**, 99 (1973).
22. J. K. G. WATSON. *J. Mol. Spectrosc.* **80**, 411 (1980).
23. H. S. PEISER, N. E. HOLDEN, P. DE BIEVRE, I. L. BARNES, R. HAGEMANN, J. R. DE LAETER, T. J. MURPHY, E. ROTH, M. SHIMA, and H. G. THODE. *Pure Appl. Chem.* **56**, 695 (1984).
24. W. GORDY and R. L. COOK. *Microwave molecular spectra*. Wiley, New York, NY. 1984.