the presence of loosely bound neutrons as might be inferred from the sequences $^{16}O+^{144,154}Sm$, $^{40}Ar+^{58,60,62,64}Ni$, and $^{86}Kr+^{58,62,64}Ni$, or to the nuclear stiffness which is believed to change appreciably within the series $^{12}C+Pd$, Sn, Te, and $^{40}Ar+Pd$, Sn, Te when crossing the Z=50 shell closure. However, the intrinsic nuclear structure of the colliding nuclei does not drastically offset the general increase of the near-barrier cross section. Moreover, its relative significance seems to diminish the heavier the systems. The data for ^{35}Cl , $^{40}Ca+^{58}\cdots ^{54}Ni$ and $^{40}Ar+^{144,148,154}Sm$ (Refs. 1 and 3), if analyzed in a similar way, would not invalidate the present systematics.

In conclusion, we find that heavy-ion fusion in the near-barrier regime by far exceeds the prediction of the simple concept of a potential barrier which depends on the separation coordinate only. The excess shows a global increase for heavier systems. The reduced radius of the coalescing system, reflecting the size of the opposing surfaces, provides a good scaling¹⁸ to this phenomenon thus indicating shape distortion or neck formation.

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Detection of the Infrared Fundamental Band of HeH+

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Nine vibration-rotation transitions of the fundamental band of HeH^+ have been observed in absorption by using a tunable, infrared, difference-frequency laser source. The molecular constants for the v=0 and v=1 levels of the $X^{1}\Sigma^{+}$ state have been determined and they agree very will with those derived from the *ab initio* calculation of Bishop and Cheung.

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Recently Oka¹ observed the vibration-rotation spectrum of H_3^+ using a tunable infrared source made by mixing radiation from an argon-ion laser and a dye laser in a LiNbO₃ crystal.² It was suggested by Oka that other ions might be detected with similar methods. We would like to report the observation of nine lines, P(4)(J=3-4) to R(4)(J=5-4), of the fundamental band (v=1-0)

of HeH+.

 ${
m HeH^+}$ is the simplest closed-shell heteronuclear molecule and is isoelectronic with ${
m H_2}$. Because the two most abundant elements in the universe are hydrogen and helium, it is not surprising that ${
m HeH^+}$ has been subjected to intense theoretical and considerable experimental scrutiny. ${
m ^{3}}$ ${
m ^4}$

Hogness and Lunn⁵ discovered HeH⁺ in 1924

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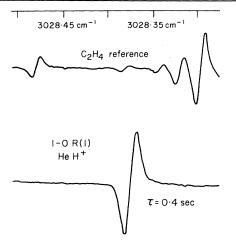


FIG. 1. The experimental trace of the R(1) line of the 1-0 infrared band of HeH⁺ at 3028.3750 cm⁻¹. The time constant was 0.4 sec.

with a mass spectrometer. Until the work of Tolliver, Kyrala, and Wing³ in 1979, however, no high-resolution spectroscopic data were available. They used Doppler-tuned ion-beam spectroscopy with a CO laser to detect the P(12) and P(13) lines of the fundamental vibration-rotation band as well as three lines of the 2-1 band. Since then they have observed more lines in higher members of the same ($\Delta v = 1$) sequence. Unfortunately, because there are no CO laser lines in the appropriate spectral region, the astrophysically important R(0) and P(1) lines of the fundamental band could not be observed. In fact, because good spectroscopic constants cannot be extracted from only two high-JP-branch lines, R(0) and P(1) are not

accurately predicted by their experimental results.

Our experimental setup was essentially identical to that of Oka.¹ The infrared radiation was generated as described for ${\rm H_3}^+$ with the exception that a Coherent 699-21 ring dye laser was used instead of a Spectra-Physics 580A laser. This provided an increase in infrared power (about a factor of 3) and, more importantly, an increase in the absolute accuracy with which transition frequencies can be measured (now about $\pm 0.001-0.002~{\rm cm}^{-1}$). The reference gases used for frequency calibration were ${\rm C_2H_4}, {\rm ^7H_2CO}, {\rm ^7}$ and ${\rm N_2O}.^{\rm 8}$ The infrared source was frequency modulated at 2.5 kHz with an amplitude of about 400 MHz.

We used a liquid-nitrogen-cooled multiple-reflection discharge cell¹ with a mixture of about 100:1 of He to H₂. The HeH⁺ absorption signal was very sensitive to the exact mixing ratio, which was optimized experimentally. This sensitivity is presumably caused by the very rapid reaction⁹

$$HeH^{+} + H_{2} \rightarrow He + H_{3}^{+}$$
,

which removes the HeH⁺ at small He-to-H₂ mixing ratios. HeH⁺ is probably produced by the reactions¹⁰

$$\text{He*} + \text{H}_2 \rightarrow \text{HeH*} + \text{H} + e$$
 and and 11

$$H_2^+(v \ge 3) + He - HeH^+ + H.$$

Our total pressure was 0.45 Torr and the optimum current density was about 30 mA/cm^2 . The path length was 32 m and the maximum absorption [for

TABLE I. Observed and calculated wave numbers of the $\operatorname{HeH}^+ v = 1 \leftarrow 0$ band (in inverse centimeters). The numbers in parentheses next to the observed column are the obs.—calc. differences obtained by use of the spectroscopic constants of Table II.

Transition	Obs.	Ab initio calc. a	Obs Calc.
R (4)	3157.2967(-1)	3157.632	-0.335
R(3)	3121.0765(1)	3121.417	-0.341
R(2)	3077.9919(4)	3078.332	-0.340
R(1)	3028.3750(-8)	3028.715	-0.340
R(0)	2972.5732(0)	2972.913	-0.340
P(1)	2843.9035(13)	2844.242	-0.339
P(2)	2771.8059(-9)	2772,140	- 0.334
P(3)	2695.0500(-5)	2695.381	-0.331
P(4)	2614.0295(4)	2614.353	-0.324
P(12) ^b	1855.905(0)	1856.152	-0.247
P (13) b	1751.971(0)	1752.198	-0.227

^a From Ref. 4.

^bFrom Ref. 3.

TABLE II.	Observed and calcula	ated spectroscopic	constants (in	n inverse centime-
ters). $1-\sigma$ und	ertainty in the last o	digits is given in p	arentheses.	

	v = 0		v = 1	
	Obs.	Cale. ^a	Obs.	Calc.a
B_{v}	33.55841(21)	33.559 161 (75)	30.839 91(18)	30.840 008 (43)
$O_v \times 10^2$	1.621 03(122)	1.620 29 (15)	1.58694(103)	1.584133(84)
$q_{\nu} \times 10^6$	5.81(15)	5.840(10)	5.76(20)	5.3736(56)
$L_n \times 10^9$	• • •	-2.870(21)	•••	- 3.218(12)
T_v	0.0	-0.00083(95)	2910.956 81 (64)	2911.297 03 (54)

^aCalculated from the *ab initio* term values of Ref. 4 for $0 \le J \le 15$.

R(1)] was about 4%. This leads to an estimate of about 10^9 molecules/cm³ for the density of HeH⁺ using the *ab initio* dipole moment function. The relative strength of the rotational lines suggests a rotational temperature of about 200 K while the linewidths are consistent with a translational temperature of about 200 K.

Figure 1 is an experimental trace of the R(1) line. The nine observed transitions of the v=1 -0 band of the $X^1\Sigma^+$ state of HeH $^+$ are presented in Table I together with the two transitions observed by Tolliver, Kyrala, and Wing. 3 The lines were fitted by using the power-series expression for vibration-rotation term values of $^1\Sigma$ molecules:

$$F_{v}(J) = T_{v} + B_{v}J(J+1) - D_{v}[J(J+1)]^{2} + H_{v}[J(J+1)]^{3} + L_{v}[J(J+1)]^{4}.$$
(1)

The values of these parameters are given in Table II. It was necessary to include terms up to H and the standard deviation of the fit was 0.0005 cm⁻¹.

The *ab initio* calculated transition frequencies shown in Table I were computed from Bishop and Cheung's v=0 and v=1 term values.⁴ They included adiabatic corrections to the Born-Oppenheimer potential in their calculation. Our transition frequencies are about 0.33 cm⁻¹ lower than their predictions as shown in the last column of Table I. This is excellent agreement (1 part in 10^4) and well within their modest estimate of 1 cm⁻¹ accuracy.

In order to compare the constants of Table II with theoretical predictions,⁴ the term values for $0 \le J \le 15$ for v = 0 and 1 were fitted to Eq. (1). The agreement for all constants, except the band origin, is within 2 or 3 standard deviations (see Table II).

One of the reasons HeH⁺ is of interest is the possibility that it might be found in extraterrestrial sources.¹² The first allowed electronic tran-

sition $A^1\Sigma^+-X^1\Sigma^+$ in the vacuum ultraviolet is expected to be very difficult to observe. Thus a high-resolution infrared search for R(0) of the fundamental band seems more promising. There is considerable controversy about the detectability of HeH⁺. However, the recent calculations of Roberge and Dalgarno¹⁴ suggest that HeH⁺ can be detected in dense clouds subjected to x-ray and extreme ultraviolet radiation and in planetary nebulae.

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